

Towards Sustainable Aviation Fuels via Electrochemical CO₂ Reduction

Andrijana Minic, Ifan E.L. Stephens, Maria-Magdalena Titirici

IMPERIAL

Aim: Develop Power to X (P2X) technology to convert CO₂ (directly captured from air) with renewable electricity to create energy-dense long-chain hydrocarbons for use in aviation and shipping.

Towards More Sustainable Energy Systems

- 73.2% of global greenhouse gas emissions are linked to energy use, mainly from fossil fuels (2.5% from aviation fuels)^[1]
- Electrochemical CO₂ reduction to sustainably produce value-added products
- CO₂ electro-reduction to energy-dense long-chain hydrocarbons as sustainable “drop-in” replacement to fossil-derived jet fuel

Research Gaps

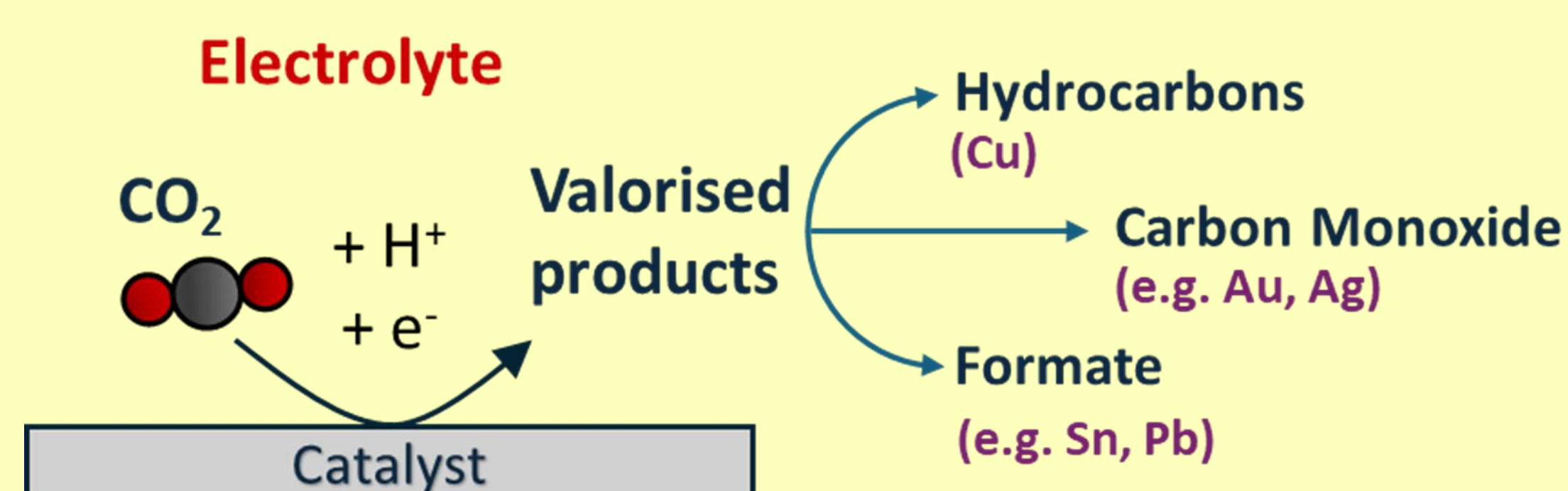


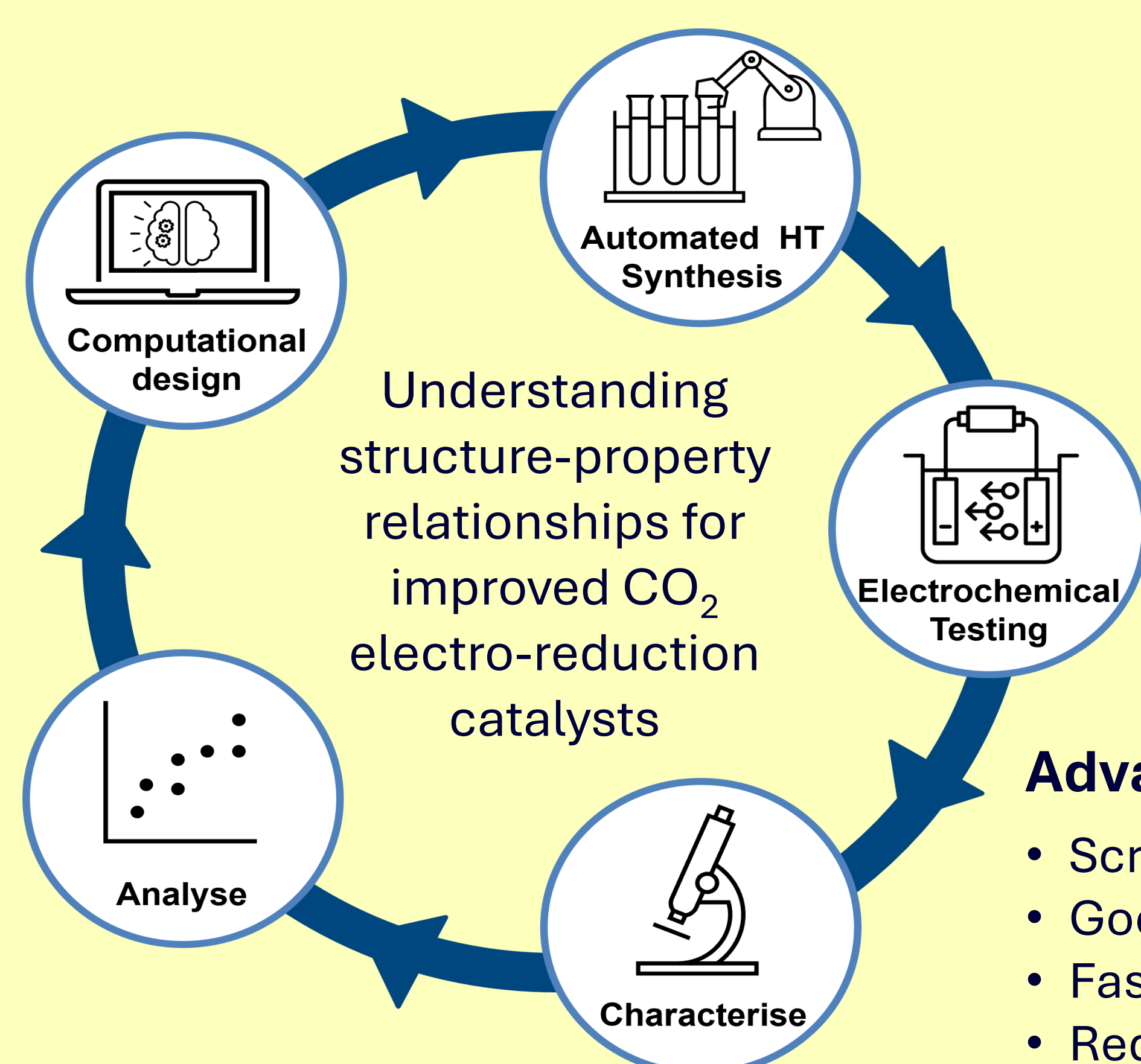
Figure 2: CO₂ electro-reduction to various catalyst-dependent product pathways.

Research Aims and Objectives

1. Determine structure-property relationships of CO₂ electrocatalysts in non-aqueous electrolytes for enhanced C-C coupling
2. Designing catalysts for enhanced C-C coupling in non-aqueous electrolytes which can provide optimised performance via controlling the hydrogenation rate and kinetics
3. Developing a High Throughput workflow for facile electrocatalyst design (and building of a data library)

Research Approach

- Using Automated High Throughput (HT) workflow



Advantages:

- Screen larger parameter space
- Good reproducibility
- Fast result accumulation
- Reduced manual labour
- More cost effective

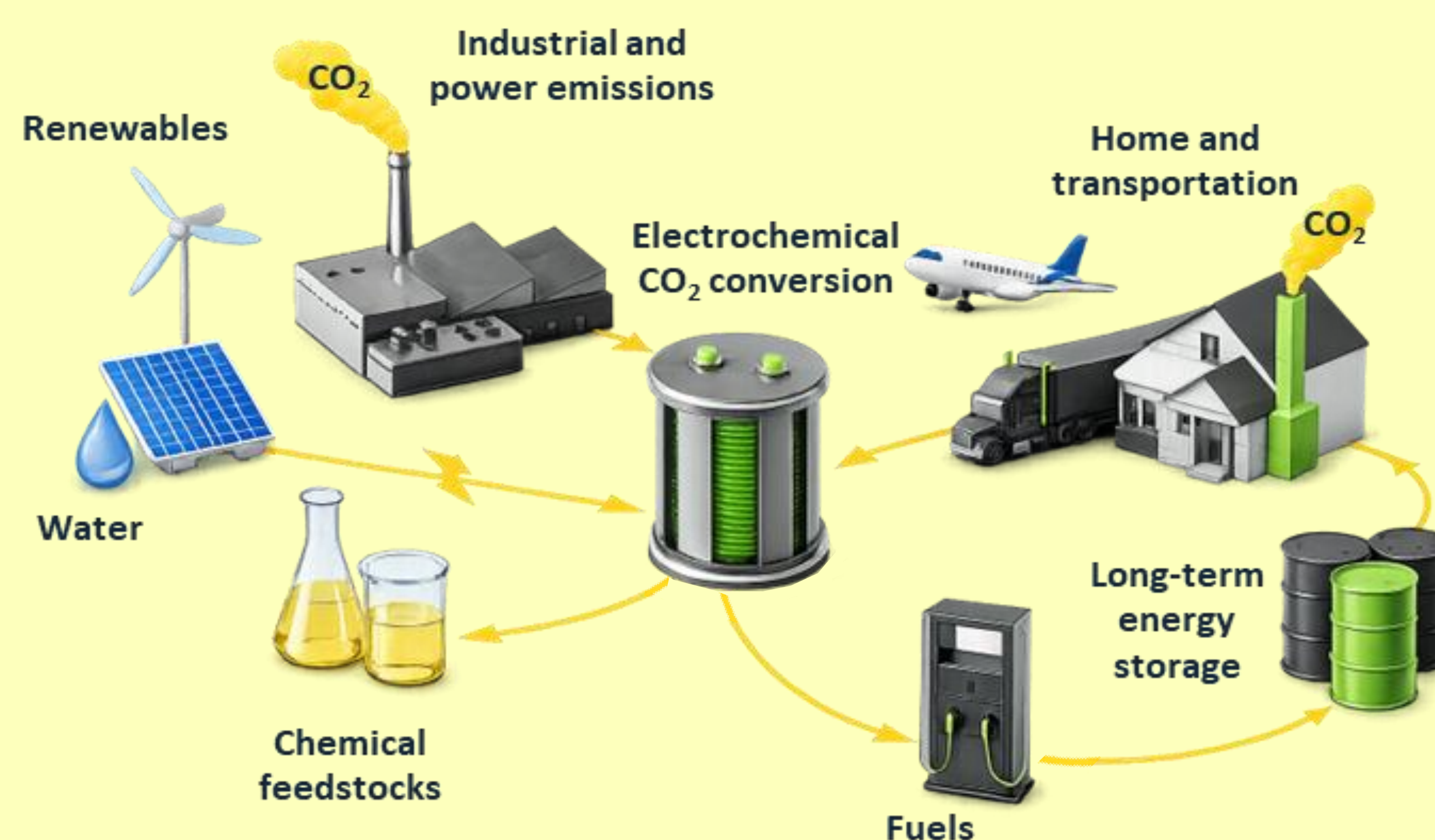


Figure 1: Electrochemical CO₂ technology integrated in an industrial system for fuels and chemical feedstocks synthesis.

- Copper (Cu) is the only catalyst to produce highly reduced species^[2]
- Most of studies have been restricted to aqueous 25°C, 1 atm studies
- Challenges of aqueous CO₂ electro-reduction: low selectivity and electrolyte carbonation
- Aprotic solvents enhance selectivity and control reaction pathways by modulating solvation, cation interactions and proton availability^[3]

Using optimised catalysts in non-aqueous (aprotic) solvent-based electrolytes

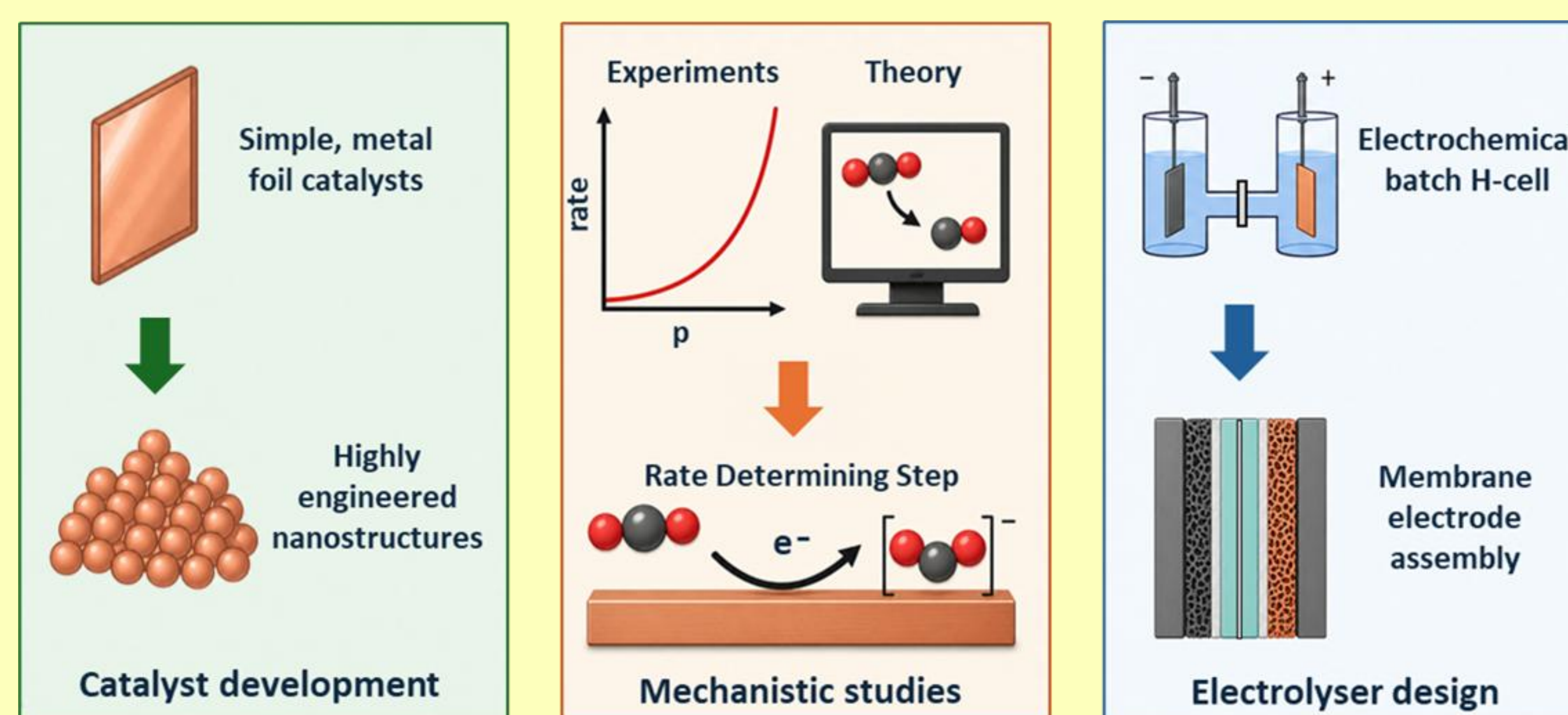


Figure 3: Overview of research areas.

Output

- A new integrated reactor combining CO₂ reduction reaction (CO₂RR) and H₂ oxidation reaction (HOR) for optimised C-C coupling towards multi-carbon products which can be blended into jet fuels

Impacts

- Mitigation of CO₂ emissions
- Storage of renewable electricity by converting surplus energy from intermittent sources into chemical energy
- More sustainable, low-carbon chemical industry that aligns with the principles of green chemistry
- Ensuring equitable and widespread access to the resulting products



Project relation to UN Sustainable Development Goals



(1) International Energy Agency, Net Zero by 2050 - A Roadmap for the Global Energy Sector, (2021)
 (2) A. Bagger, et al., Electrochemical CO₂ Reduction: A Classification Problem, *ChemPhysChem* 18, 3266–3273, (2017)
 (3) A. Chu and Y. Surendranath Aprotic Solvent Exposes an Altered Mechanism for Copper Catalyzed Ethylene Electrosynthesis, *J. Am. Chem. Soc.* 144, 5359–5365, (2022)